

# Cobalt Oxide nanomaterial over MoS<sub>2</sub> quantum dots as a heterojunction photocatalyst for water splitting.

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## Abstract:

Modern technological era demands for energy, and we are running short of conventional sources. Newer ways to support the energy demand is to be discovered. Hydrogen is a promising energy source with the use in local commuting vehicles (HICEV), liquid propellant rockets, and fuels cells. Until 2022 approximately 95% hydrogen is produced from steam reforming of fossil fuels or oxidation of methane [1].

In our study we have designed a Surface Plasmon Resonance (SPR) enhanced cobalt tetraoxide (Co<sub>3</sub>O<sub>4</sub>) photocatalyst to achieve water splitting reaction to produce hydrogen and oxygen. Cobalt tetraoxide has been well established in the semiconductor industry and is actively researched as a photocatalyst due to its enormous active sites, stability, and band gap tunability by tuning the synthesis process [2]. Our research focuses on synthesising Cobalt tetraoxide using a rapid, microwave irradiation technique.

We also introduced SPR effect on the Co<sub>3</sub>O<sub>4</sub> by incorporating Molybdenum disulfate quantum dots (MoS<sub>2</sub>QD) to enhance Cobalt tetraoxide light absorption in order to further improve the water splitting efficiency. Finally, the SPR-Co<sub>3</sub>O<sub>4</sub> will be integrated with a microfiltration membrane for continuous filtration and photocatalysis to further analyse the performance of Co<sub>3</sub>O<sub>4</sub> standalone and SPR induced Co<sub>3</sub>O<sub>4</sub>/ MoS<sub>2</sub>QD using gas chromatograph for Hydrogen and Oxygen detection.

In fig. 1a, the reducer concentration (PVA and NaOH) affects the light absorbance for a fixed synthesis temperature. The reason to this is the change in morphology of the Cobalt tetraoxide. Fig. 1 (b) also demonstrates that, keeping reducer concentration same but change in synthesis temperature affects the light absorbance. 200°C-1,1.5 PVA is observed optimal synthesis temperature for Co<sub>3</sub>O<sub>4</sub>. So, both reducer and temperature is governing parameters for Co<sub>3</sub>O<sub>4</sub> performance.

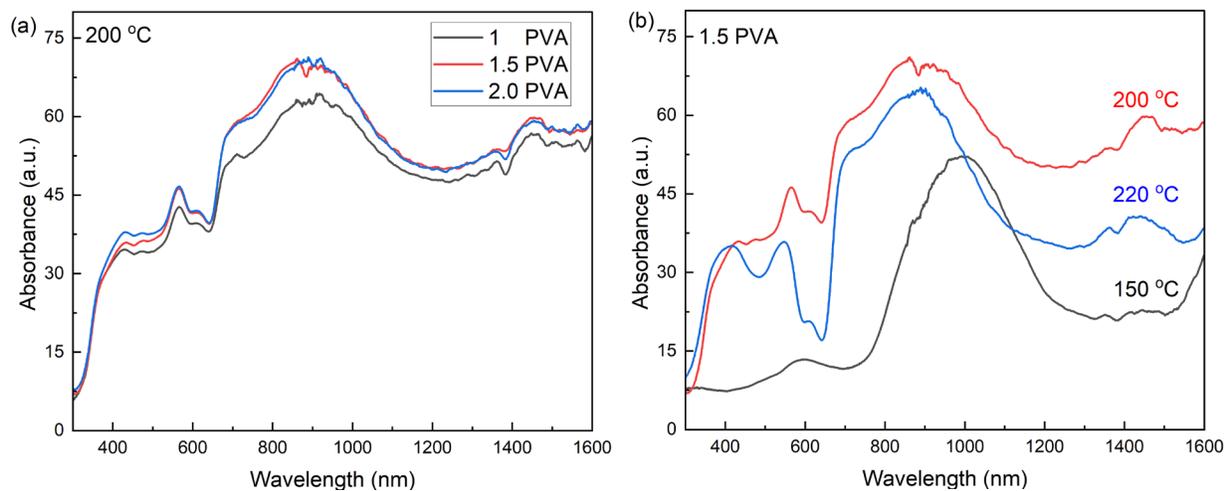


Fig. 1 (a) UV-Vis-NIR spectra for dif. 1.5 PVA and constant temp

Fig. 1 (b) UV-Vis-NIR spectra for 1.5 PVA dif. Synthesis temp.

Keywords: Water splitting, Hydrogen, Oxygen, Cobalt Oxide Photocatalyst.

## Reference

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